Annulated Heterocycles

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Rh-Catalyzed Transannulation of Pyridotriazoles with Alkynes and Nitriles**

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Transition-metal-catalyzed annulations are widely used in the synthesis of heterocyclic compounds. One of the most efficient methods for the construction of five-membered oxygen-containing heterocycles involves the annulation of diazocarbonyl compounds with alkynes and nitriles. Thus, Davies et al. and Padwa et al. have employed this method for the synthesis of furans (X = CH), and Helquist et al. for the preparation of oxazoles (X = N) [Eq. (1)]. In contrast, analogous transformations of α -imino diazo compounds, which may lead to the formation of pyrole and imidazole rings, are unknown. Herein we report an efficient, direct, Rh-catalyzed transannulation of pyridotriazoles with alkynes and nitriles that leads to indolizines (X = CH) and imidazopyridines (X = N), respectively [Eq. (2)].

$$R^{1} \xrightarrow{N_{2}} + R^{3} = X \xrightarrow{Rh^{||}} R^{1} \xrightarrow{R^{2}} X$$

$$X = CH \text{ (Davies, Padwa)}$$

$$X = N \text{ (Helquist)}$$

It has been shown that 2-pyridyl diazo compounds 1^[6] transform into their cyclic triazole form 2^[7] upon storage [Eq. (3)], and it is also known that some of these cyclic triazoles can still undergo transformations that are characteristic of diazo compounds.^[8] This phenomenon has been attributed to the closed/open form equilibrium of N-fused triazoles in solution,^[9] which can produce trace to significant

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amounts of **1**. The position of this equilibrium depends on the temperature and the substitution pattern of the triazole. [9b] Thus, introduction of a halogen substituent at C7 ($R^1 = Cl$) shifts the equilibrium to the left, which has been explained in terms of nonbonding repulsion between the lone pair of the halogen and that of the nitrogen in the *peri* position. [10]

To evaluate the feasibility of using triazoles as precursors of Rh carbenoids we investigated the reaction of triazoles **3a** and **3b** with triethylsilane in the presence of a catalytic amount of rhodium(II) acetate, which is a method developed by Doyle and coworkers^[11] for the efficient trapping of Rh carbenoids [Eq. (4)]. Not surprisingly, pyridotriazoles **3a** and **3b** behave differently under these reaction conditions. Thus, while the 7-H derivative **3a** remains unaffected, the 7-chlorosubstituted compound **3b** is smoothly converted into **4**, which is the product of carbenoid insertion into the Si–H bond. These experiments clearly indicate that 7-halo-substituted pyridotriazoles can indeed serve as convenient precursors of Rh carbenoids.

Next, to test our hypothesis regarding the annulation of α -imino diazo compounds with alkynes to form a pyrrole ring, we treated triazole $\bf 3b$ with phenylacetylene in the presence of rhodium(II) acetate. This reaction proceeded smoothly to produce a mixture of cyclopropene $\bf 5$ and indolizine $\bf 6a$ with yields of $\bf 68\%$ and $\bf 28\%$ of isolated product, respectively [Eq. (5)]. Surprisingly, cyclopropene $\bf 5$ does not undergo further isomerization into indolizine $\bf 6a$ under these reaction conditions. [12] The ratio of these products remained constant throughout the course of the reaction, thereby suggesting an independent path for the formation of $\bf 6a$.

We found, however, that the selectivity of the transannulation (6 over 5) could be dramatically improved by using

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rhodium(II) heptafluorobutyrate as catalyst.^[13] Thus, transannulation of **3b** with a series of aryl and alkenyl alkynes^[14] proceeded highly chemoselectively (90:10 to 95:5 vs. cyclopropene) to produce indolizines **6**^[15] in good yields (Table 1). Electron-rich, electron-deficient, and sterically hindered aryl alkynes were nearly equally effective in this reaction.

Table 1: Rhodium(II)-catalyzed transannulation of triazole **3b** with alkynes.

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Entry	Alkyne	Product	Yield [%] ^[a]			
1	<u></u>	6 a	78			
2	H ₃ C	6 b	80			
3	CH ₃	6 c	73			
4	MeO-	6 d	85			
5		6 e	70			
6		6 f	65			
7	MeO	6 g	57			

[a] Yield of isolated product. Indolizines **6** were accompanied by 5–10% of the corresponding cyclopropenes **5**; these compounds were readily separable by column chromatography.

Inspired by the successful formation of an N-fused pyrrole ring from the transannulation of triazoles with alkynes, we examined the formation of an N-fused imidazole ring in the reaction of 3 with nitriles and found that pyridotriazoles 3 react smoothly with a variety of aryl, alkyl, and alkenyl nitriles in the presence of $Rh_2(OAc)_4$ (1 mol%) in toluene at 60°C (Table 2) to afford N-fused imidazopyridines 7 in reasonable to high yields.

Both 3-carbomethoxy- (Table 2, entries 1–9) and 3-aryl-(Table 2, entry 10) pyridotriazoles are equally efficient in this reaction. Moreover, 7-bromo- (Table 2, entry 11) and even 7-methoxy-substituted (Table 2, entry 12) triazoles proved to be good substrates for this transannulation reaction.

We propose the following mechanism for this novel Rh-catalyzed transformation (Scheme 1). First, pyridotriazole 3 undergoes closed/open form equilibrium^[9] to produce small amounts of diazo compound 1 which, upon reaction with rhodium(II) carboxylate, generates the Rh-carbenoid species I. A direct nucleophilic attack^[18] of alkyne or nitrile 8 on species I produces ylide species II, according to path A, which

Table 2: Rhodium(II)-catalyzed transannulation of triazoles with nitriles.

Entry	R ¹	R ²	Triazole	R³	Product	Yield [%] ^[a]
1	Cl	CO ₂ Me	3 b	p-Tol	7 a	89
2	Cl	CO ₂ Me	3 b	Ph	7 b	83
3	Cl	CO ₂ Me	3 b	p-Me(O)CC ₆ H ₄	7 c	54
4	Cl	CO ₂ Me	3 b	Bn	7 d	63
5	Cl	CO_2Me	3 b	<i>n</i> Pr	7 e	75
6	Cl	CO ₂ Me	3 b	<i>c</i> Pr	7 f	74
7	Cl	CO_2Me	3 b	<i>t</i> Bu	7 g	69
8	Cl	CO ₂ Me	3 b	<u></u> -ξ	7 h	66
9	Cl	CO ₂ Me	3 b	CH₂SiMe₃	7 i	70
10	Cl	p-CF ₃ C ₆ H ₄	3 c	Ph	7 j	82
11	Br	p-CF ₃ C ₆ H ₄	3 d	<i>p</i> -Tol	7 k	73
12	OMe	p-CF ₃ C ₆ H ₄	3 e	<i>n</i> Pr	71	51

[a] Yield of isolated product.

Scheme 1. Plausible mechanisms for the Rh-catalyzed transannulation of pyridotriazoles with alkynes and nitriles. Y = N, CR''.

then cyclizes to form **6** or **7** via cyclic zwitterion **III**. Alternatively (path B), [2+2] cycloaddition of **I** and **8** leads to metallacyclobutene **IV**, which can also be formed by cyclization of **II**. [19] Rhodacycle **IV** then undergoes metathesis

to produce Rh carbenoid **V** which, upon 6π -electrocyclization and subsequent reductive elimination, furnishes product **6** or **7**. [2+1] Cycloaddition of **I** with **8** (path C) accounts for the formation of cyclopropene **5** in the presence of rhodium(II) acetate [see Eq. (5)]. As discussed above, **5** does not transform into heterocycle **6** under these reaction conditions. [12]

In summary, we have developed an efficient Rh-catalyzed transannulation of pyridotriazoles for the formation of pyrrolo- and imidazopyridines, which are important fused heterocyclic scaffolds.^[20] We have also demonstrated that some of these pyridotriazoles can serve as stable^[13] and convenient^[21] precursors of Rh carbenoids.

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- [1] For recent reviews, see: a) I. Nakamura, Y. Yamamoto, *Chem. Rev.* **2004**, *104*, 2127; b) G. Zeni, R. C. Larock, *Chem. Rev.* **2004**, *104*, 2285; c) M. Rubin, A. W. Sromek, V. Gevorgyan, *Synlett* **2003**, 2265.
- [2] a) H. M. L. Davies, K. R. Romines, *Tetrahedron* 1988, 44, 3343;
 b) H. M. L. Davies, W. R. Cantrell, Jr., K. R. Romines, J. S. Baum, *Org. Synth.* 1992, 70, 93.
- [3] a) F. R. Kinder, A. Padwa, Tetrahedron Lett. 1990, 31, 6835;
 b) A. Padwa, J. M. Kassir, S. L. Xu, J. Org. Chem. 1991, 56, 6971;
 c) A. Padwa, F. R. Kinder, J. Org. Chem. 1993, 58, 21.
- [4] For recent applications, see: a) V. Gettwert, F. Krebs, G. Maas, Eur. J. Org. Chem. 1999, 1213; b) C. Batsila, G. Kostakis, L. P. Hadjiarapoglou, Tetrahedron Lett. 2002, 43, 5997.
- [5] R. Connell, F. Scavo, P. Helquist, Tetrahedron Lett. 1986, 27, 5559.
- [6] For cyclopropanation with 2-pyridyl diazo compounds, see: H. M. L. Davies, R. J. Townsend, J. Org. Chem. 2001, 66, 6595.
- [7] M. Regitz, Angew. Chem. 1967, 79, 786; Angew. Chem. Int. Ed. Engl. 1967, 6, 733.
- [8] See, for example: B. Abarca-González, *J. Enzym. Inhib. Med. Chem.* **2002**, *17*, 359, and references therein.
- [9] a) M. Regitz, B. Arnold, D. Danion, H. Schubert, G. Fusser, Bull. Soc. Chim. Belg. 1981, 90, 615; b) G. L'abbé, Bull. Soc. Chim.

- Belg. 1990, 99, 281; c) G. L'abbé, F. Godts, S. Toppet, J. Chem. Soc. Chem. Commun. 1985, 589; d) G. L'abbé, I. Luyten, S. Toppet, J. Heterocycl. Chem. 1992, 29, 713; e) G. L'abbé, F. Godts, S. Toppet, Bull. Soc. Chim. Belg. 1986, 95, 679.
- [10] B. Abarca-González, R. Ballesteros, F. Mojarred, G. Jones, D. J. Mouat, J. Chem. Soc. Perkin Trans. 1 1987, 1865.
- [11] V. Bagheri, M. P. Doyle, J. Taunton, E. E. Claxton, J. Org. Chem. 1988, 53, 6158.
- [12] For a related transformation of 3-carbonyl cyclopropenes into furans, see refs. [2a] and [3a].
- [13] See the Supporting Information for details.
- [14] The use of aliphatic alkynes resulted in sluggish and incomplete reactions.
- [15] Halo-substituted N-fused heterocycles can be further functionalized by cross-coupling reactions.^[16] We have also found that the chloride in 6a can be quantitatively removed by treatment with PdCl₂/HSiEt₃.^[17]
- [16] For a review of coupling reactions of aryl chlorides, see: A. F. Littke, G. C. Fu, Angew. Chem. 2002, 114, 4350; Angew. Chem. Int. Ed. 2002, 41, 4176.
- [17] R. Boukherroub, C. Chatgilialoglu, G. Manuel, Organometallics 1996, 15, 1508.
- [18] A. Padwa, D. J. Austin, A. T. Price, M. A. Semones, M. P. Doyle, M. N. Protopopova, W. R. Winchester, A. Tran, *J. Am. Chem. Soc.* 1993, 115, 8669.
- [19] T. R. Hoye, C. J. Dinsmore, D. S. Johnson, P. F. Korkowski, J. Org. Chem. 1990, 55, 4518.
- [20] For selected reports of the biological activity of indolizines, see: a) S. Hagishita, M. Yamada, K. Shirahase, T. Okada, Y. Murakami, Y. Ito, T. Matsuura, M. Wada, T. Kato, M. Ueno, Y. Chikazawa, K. Yamada, T. Ono, I. Teshirogi, M. Ohtani, J. Med. Chem. 1996, 39, 3636; b) L.-L. Gundersen, K. E. Malterud, A. H. Negussie, F. Rise, S. Teklu, O. B. Østby, Bioorg. Med. Chem. 2003, 11, 5409; for reports on the biological activity of imidazopyridines, see: c) D. Kim, L. Wang, J. J. Hale, C. L. Lynch, R. J. Budhu, M. MacCoss, S. G. Mills, L. Malkowitz, S. L. Gould, J. A. DeMartino, M. S. Springer, D. Hazuda, M. Miller, J. Kessler, R. C. Hrin, G. Carver, A. Carella, K. Henry, J. Lineberger, W. A. Schleif, E. A. Emini, Bioorg. Med. Chem. Lett. 2005, 15, 2129; d) S. Nakahara, A. Kubo, Y. Mikami, J. Ito, Heterocycles 2006, 68, 515.
- [21] The use of 3 does not require special slow-addition techniques as the concentration of 1 in the reaction mixture is always low.^[9]

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